Lists of atomic coordinates, displacement parameters, structure factors and complete geometry have been deposited with the IUCr (Reference: AB1419). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

## References

- Agmon, I., Kaftory, M., Nelson, S. F. & Blakstock, S. C. (1986). J. Am. Chem. Soc. 108, 4477-4484.
- Allen, F. H., Kennard, O. & Taylor, R. (1983). Acc. Chem. Res. 16, 146-153.
- Brock, C. P., Demir, A. S. & Watt, D. S. (1995). Acta Cryst. C51, 2434-2437.
- Enraf-Nonius (1993). CAD-4 Express Software. Version 1.1. Enraf-Nonius, Delft, The Netherlands.
- Fair, C. K. (1990). MolEN. An Interactive Intelligent System for Crystal Structure Analysis. Enraf-Nonius, Delft, The Netherlands.
- Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.
- Spek, A. L. (1992). PLUTON92. Molecular Graphics Program. University of Utrecht, The Netherlands.

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# 4,4'-Dimethyl-2,2'-bipyridine 1,1'-Dioxide

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## Abstract

There are four molecules in a unit cell of the title compound,  $C_{12}H_{12}N_2O_2$ , and the structure of only half a molecule has been determined because of its twofold rotational symmetry. The dihedral angle between the two pyridine rings is 61.3° and the non-bonded distance between O atoms in the *N*-oxide groups is 2.96 Å. The skew conformation of the skeleton is almost the same as that of the tetrakis(2,2'-bipyridine 1,1'-dioxide)-lanthanum(III) complex.

## Comment

The bidentate ligand 2,2'-bipyridine 1,1'-dioxide (bpdo) forms a skewed seven-membered ring with a metal ion. Since the skew conformation produces a pair of enantiomers, a tris(bpdo) complex has four possible racemic pairs of conformational isomers. The tris-(bpdo)– $Cr^{III}$  complex, however, crystallizes in one of the four isomers and then isomerizes and racemizes on dissolution in water (Kanno & Fujita, 1987). This

© 1997 International Union of Crystallography Printed in Great Britain – all rights reserved indicates that the skewed bpdo-chelate ring is flexible and changes its conformation rather easily in solution. The dihedral angle between the pyridine rings, which seems to represent a distortion of the seven-membered ring, is  $61.4^{\circ}$  in the La<sup>III</sup> complex (Al-Karaghouli, Day & Wood, 1978) and  $53.7^{\circ}$  in the U<sup>IV</sup> complex (Alcock & Roberts, 1987). The present study was undertaken in order to estimate the dihedral angle between the pyridine rings of the bpdo molecule with methyl groups at the 4,4'-positions, (I).



The title molecule has a twofold rotational axis and the parameters of half of the molecule have been determined (Fig. 1). The pyridine ring is planar within 0.03 Å, but the O atom of the *N*-oxide group is slightly out of the plane (0.13 Å), away from the other O atom



Fig. 1. ORTEP drawing (Johnson, 1965) of the whole title molecule, half of which is an asymmetric unit, showing the atomic numbering scheme. Displacement ellipsoids of the non-H atoms are shown at the 50% probability level. H atoms are shown as spheres of arbitrary radii.



Fig. 2. The crystal structure of (I) viewed along the b axis.

(Fig. 2). The dihedral angle between the pyridine rings is  $61.3^{\circ}$ , in good agreement with the value for the La<sup>III</sup> complex (Al-Karaghouli, Day & Wood, 1978). The central C—C distance is also similar to that in the La<sup>III</sup> complex, but the N—O distance is shorter by 0.03 Å. There are no unusual short intermolecular distances.

## Experimental

The title compound was prepared according to the method of Kanno, Yamamoto, Murahashi, Utsuno & Fujita (1991) and recrystallized from methanol.

#### Crystal data

C <sub>12</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub> $M_r = 216.24$ Monoclinic C2/c a = 11.232 (4) Å b = 9.175 (3) Å c = 10.444 (4) Å $\beta = 107.09$ (3)° V = 1028.8 (7) Å <sup>3</sup> Z = 4 $D_x = 1.396$ Mg m <sup>-3</sup> $D_m = 1.38$ Mg m <sup>-3</sup>	Mo $K\alpha$ radiation $\lambda = 0.71073$ Å Cell parameters from 23 reflections $\theta = 14.3-17.5^{\circ}$ $\mu = 0.097 \text{ mm}^{-1}$ T = 298  K Needle $0.40 \times 0.30 \times 0.25 \text{ mm}$ Colourless
a mixture of benzene and carbon tetrachloride	
Data collection	
MAC Science MXC3 diffractometer $2\theta/\omega$ scans Absorption correction: none 1328 measured reflections 1087 independent reflections 948 reflections with $l > 1.25\sigma(l)$	$R_{int} = 0.027$ $\theta_{max} = 26.43^{\circ}$ $h = -14 \rightarrow 13$ $k = 0 \rightarrow 11$ $l = 0 \rightarrow 13$ 3 standard reflections every 100 reflections intensity decay: < 2%

#### Refinement

Refinement on F $(\Delta/\sigma)_{max} = 0.0025$ R = 0.0572 $\Delta\rho_{max} = 0.26$  e Å<sup>-3</sup>wR = 0.0585 $\Delta\rho_{min} = -0.29$  e Å<sup>-3</sup>S = 1.590Extinction correction: none948 reflectionsScattering factors from Inter-<br/>national Tables for X-ray $T_3$  parametersnational Tables for X-ray<br/>Crystallography (Vol. IV) $w = 1/[\sigma^2(F_o) + 0.0005F_o^2]$ 

## Table 1. Selected geometric parameters (Å, °)

01N1 N1C1 N1C5 C1C1 <sup>i</sup> C1C2	1.301 (2) 1.370 (2) 1.362 (2) I.482 (2) 1.375 (2)	C2—C3 C3—C4 C3—C6 C4—C5	1.391 (2) 1.387 (2) 1.494 (3) 1.361 (2)
$\begin{array}{c} 01 - N1 - C1 \\ 01 - N1 - C5 \\ C1 - N1 - C5 \\ N1 - C1 - C1^{1} \\ N1 - C1 - C2 \\ C1^{1} - C1 - C2 \end{array}$	120.8 (2)	C1C2C3	122.1 (2)
	120.4 (2)	C2C3C4	115.7 (2)
	118.8 (2)	C2- C3C6	121.4 (2)
	117.3 (2)	C4C3C6	122.8 (2)
	120.0 (2)	C3C4C5	122.1 (2)
	122.6 (2)	N1C5C4	121.1 (2)

Symmetry code: (i) -x, y,  $\frac{1}{2} - z$ .

The space group was finally determined from the Wilson plot. The structure was solved by direct methods and refined by fullmatrix least-squares calculations using *CRYSTAN* (Gilmore & Brown, 1988). H atoms were found from the difference Fourier map at the final stage and were restricted to their parent atoms with C—H distances of 0.96 Å and the corresponding equivalent isotropic displacement parameters.

Data collection: MXC software (MAC Science Co. Ltd, 1989). Cell refinement: MXC software. Data reduction: CRYS-TAN. Program(s) used to solve structure: CRYSTAN. Molecular graphics: ORTEP (Johnson, 1965) and CRYSTAN. Software used to prepare material for publication: CRYSTAN.

Lists of atomic coordinates, displacement parameters, structure factors and complete geometry have been deposited with the IUCr (Reference: OH1103). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

#### References

- Alcock, N. W. & Roberts, M. M. (1987). Acta Cryst. C43, 476–478.Al-Karaghouli, A. R., Day, R. O. & Wood, J. S. (1978). Inorg. Chem.17, 3702–3706.
- Gilmore, C. J. & Brown, S. R. (1988). Acta Cryst. A44, 1018–1021. Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge
- National Laboratory, Tennessee, USA. Kanno, H. & Fujita, J. (1987). Bull. Chem. Soc. Jpn, 60, 589-594.
- Kanno, H., Yamamoto, J., Murahashi, S., Utsuno, S. & Fujita, J. (1991). Bull. Chem. Soc. Jpn, 64, 2936-2941.
- MAC Science Co. Ltd (1989). Operation Manual of of the MXC Four-Circle Diffractometer. MAC Science Co. Ltd, Yokohama, Japan.

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# *meso*-2,2,2',2'-Tetrachloro-1,1'-diphenyl-1,1'-bicyclopropyl and *meso*-2,2,2',2'-Tetrabromo-1,1'-diphenyl-1,1'-bicyclopropyl

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#### Abstract

Molecules of *meso*-2,2,2',2'-tetrachloro-1,1'-diphenyl-1,1'-bicyclopropyl,  $C_{18}H_{14}Cl_4$ , (1), and *meso*-2,2,2',2'tetrabromo-1,1'-diphenyl-1,1'-bicyclopropyl,  $C_{18}H_{14}Br_4$ , (2), adopt *gauche* conformations. Compound (1) produced two types of crystals upon recrystallization. The phenyl rings of the two forms of compound (1), as well as those of compound (2), are in a nearly perpendicular orientation. All three structures have unsymmetrical